

Loyola University Chicago

January 13, 2014 – May 02, 2014 **Lectures / Discussions / Exams**

Spring 2014

Syllabus: Biochemistry: CHEM 361/461; BIOL 366 Sec. 001; *Discussion* 002, 003

Lecture: T, R: 01:00 PM – 02:15 PM Galvin Auditorium, Sullivan Center

Discussion: 002 *M* 11:30 AM – 12:20 PM Dumbach 125; 003 *M* 12:35 PM - 01:25 PM Dumbach 125

Instructor: Donald May Contact: dmay4@luc.edu **Office:** Flanner Hall 403; **Hours:** **MON.:** 10:30 AM – 11:15 AM; **TUES.:** only directly before exams 11:30 AM - 12:30 PM. Other times by appointment.

Textbook: Biochemistry, Campbell/ Farrell, 7th ed., Brooks-Cole, Cengage Learning, 2012.

Method of instruction: Lecture and discussion. Lectures may be supplemented with classroom discussion, use of molecular models, use of multimedia, and/or use of computer based materials as well as individual and/or group problem solving. On-line homework, will count toward the final grade. Students must log-in and create an account with their access code: Access code: accompanied with the text or purchased on-line

<https://owl.cengage.com/owl-c/register/owlmgr.cgi?Mode=3&Server=owl-loyolaunivchicagocampbellbiochem7e&DatabaseID=8103>

<https://owl.cengage.com/owl-c/user/loginpage.cgi?Server=owl-loyolaunivchicagocampbellbiochem7e&UserType=Student>

Supplemental suggested textbook homework problems, for each chapter may be given but are not to be turned in for grading. Discussions may incorporate explanation of theory, review of homework questions, review of previous exam questions or completion of lecture material, if necessary. Graded exams will be returned in the next discussion. Issues with graded exams must be submitted within 7 days of being returned, otherwise scores will be considered final. Students must submit a signed statement requesting a review of the exam question(s), although the entire exam is now subject to being re-graded.

Grading: Semester grades will be determined by the following criteria:

Exams will incorporate theory up to and including all lectures/discussions/homework, prior to the exam.

Weekly on-line homework contributing 10% toward the final grade (with individual chapter due dates);

Three (3) in-class unit exams; Each unit exam will have 100 points possible; There will be 20-25 multiple choice questions at 3 points each; 3-5 long answer questions of varying point values.

The comprehensive final exam which will have 200 points possible and have a similar format to the unit exams.

Final grades will be determined from one of the following exam contribution options, whichever is higher:

OPTION 1: All three (3) unit exams at 20% each = 60%; final exam 30% = 90%

OPTION 2: Best two (2) unit exams at 20% each = 40%; final exam 50% = 90%

OPTION 1: On-line Homework: 10%

OPTION 2: On-line Homework: 10%

3 Unit Exams@60% + Final Exam@30%

2 Unit Exams@40% + Final Exam @50%

Total: 100%

Total: 100%

No early and no make-up in-class exams; No late homework. For a single, missed in-class unit exam, Option 2 automatically will be utilized to determine the final course grade. Any subsequent missed in-class exams will be scored as zero. The student must have a valid and verifiable reason for missing the final exam, such as an extreme emergency or serious illness requiring hospitalization, and so forth, to be eligible for a make-up final exam. A make-up final exam will be in a different format. If a verifiable and valid reason cannot be provided, a zero score for the final exam will be recorded. See attached schedule. Exam Dates (tentative):

EXAM I: Tuesday, February 11; EXAM II: Tuesday, March 18; EXAM III: Tuesday, April 15;

FINAL EXAM: Friday, May 02, 1-3 PM

Final course grade assigned: A: 100% – 87.0% A- : 86.9% - 83.0% B+: 82.9% - 79.0%

B: 78.9% - 75.0% B-: 74.9% - 71.0% C+: 70.9% - 67.0% C: 66.9% - 63.0% C-: 62.9% - 59.0%

D+: 58.9% - 55.0% D: 54.9% - 51.0% F: < 51.0%

Students are not allowed to leave during exams. If you leave, you must turn in your exam and you will be considered finished with the exam. Students cannot begin an exam and decide not to complete it. Students must turn in all exam materials when finished. Exams cannot be taken from lecture: see Academic Integrity Violations. Students must bring and present their Loyola I.D. for each exam.

Student Conduct: Only students officially enrolled may attend lectures and discussions. At all times students are expected to conduct themselves in a mature and professional manner, which includes but is not limited to: treating everyone in class with courtesy and respect, avoidance of extraneous comments and small group discussions during lecture. Eating, chewing gum/tobacco products and drinking (food items) are not allowed. Students are expected to take care of their personal/professional matters before lectures/discussions/exams. Additionally radios, headphones,

cell-phones or similar devices must be in silent mode and are not permitted during lectures/discussions/exams. If a cell phone rings (beeps, buzz, etc.) during discussions or lectures, the student will be asked to leave. Not all contingencies can be listed but inappropriate conduct will be addressed. If a cell phone rings (beeps, buzz, etc.) during any exam, the exam will be collected and the student will not be allowed to continue, since this constitutes using an outside resource. Students are expected to take care of any professional/personal issues before the exams. Students are not allowed to leave the room during exams until their exam is handed in for grading. If you leave, you must turn in your exam and you will be considered finished. Please keep noises and sounds to a minimum. When leaving, be respectful and leave quietly. During exams, only religious caps/ hats/hoods are allowed: nonreligious caps, hats, hoods, visors and so forth, will not be allowed to be worn during exams. All personal materials, besides pencils, calculators and erasers, will be put away. Other exam instructions will be given and thus it is expected that students will be on time and ready for the start of the exam. The visual or audio recording of the lectures and discussions is not allowed generally but exceptions can be made for extraordinary circumstances.

Academic Integrity: Consult the Undergraduate Studies Handbook for additional information. All students in this course are expected to have read and to abide by the demanding standard of personal honesty, drafted by the College of Arts & Sciences, which can be viewed at:

http://www.luc.edu/cas/pdfs/CAS_Academic_Integrity_Statement_December_07.pdf

For on-line homework, students creating multiple accounts will be considered in violation of academic integrity. All exams are self-contained: closed book and closed note. During exams, violations include but are not limited to: cell phone ringing, opening a book-bag or back-pack during an exam, using unauthorized notes or books, looking at another student's exam, talking to another student, taking a copy of the exam from the room and so forth. Students caught cheating will receive a zero score for the exam and this exam will not be allowed to be dropped: Option 1 above will automatically apply. Further actions will also result. Anything submitted that is incorporated as part of your grade in this course must represent your own work, unless indicated otherwise. All exams are closed book and closed note: No external materials/notes/books or personnel are allowed. During exams, violations include but are not limited to: cell phone ringing, answering/using a cell phone, using unauthorized notes or books, looking at another student's exam, talking to other students, opening and/or utilizing anything in your book bag, and so forth. Any student found to be in violation or cheating will, at minimum, be given a zero for the assignment/exam and the incident will be reported to the Chemistry Department Chair and the Office of the CAS Dean. Depending on the seriousness of the incident, additional sanctions may be imposed.

Course Practices Required:

Attending all lectures and discussions on time; College-level writing skills on exams: Communication skills for discussion and articulation of questions: Completion of homework and reading assignments. It is recommended that the student read through each chapter before lecture and eventually work through the suggested problems.

Learning Objectives:

Course introduces bio- molecule monomers, macromolecules, and processes found in living organisms. Content includes structures of amino acids, nucleotides, lipids, and sugars; corresponding macromolecular structures, i.e., proteins, nucleic acids, membranes, and polysaccharides as related to their biological functions; kinetics and mechanism of enzymatic reactions, the central metabolic pathways, the genetic code and developments in biotechnology. Students who successfully complete this course will be able to do the following, at an acceptable level (including but not limited to): Identify and describe biomolecules including carbohydrates, amino acids/proteins and nucleotide/nucleic acids, lipids/lipid bilayer constituents, Choose appropriate buffer system; calculate the ratios of weak acid to conjugate base; determine the pKa from the associated titration curve; Show the major form of an amino acid/polypeptide including the zwitterion, at different pH values; track the fate of an oxygen molecule from inhalation in the lungs, track the fate of a carbon dioxide molecule produced from the TCA cycle, identify the kinetics of an enzymatic process; identify the substrates, enzymes and products in both catabolic and anabolic metabolism;

Important Dates:

**Monday, January 20: No classes: Holiday; Tuesday, February 10: Summer Registration
March 03 – 08: No classes: Spring Break; Monday, March 24: Last day for “W” otherwise “WF”
Thursday, April 17: No classes after 04:15 PM: Holiday
Monday, April 21: Classes beginning after 04:15 PM are held; Saturday, April 26: Classes End**

Lecture Outline (tentative / subject to change)

Week	Date	Chapter	Topic
01	T 01/14 R 01/16	01, 02 02	Introduction; Water and pH Buffers; Henderson-Hasselbalch equation
02	T 01/21 R 01/23	03 04	Amino Acids and Polypeptides Protein Structure and Non-covalent Interactions
03	T 01/28 R 01/30	04 06	Protein Folding; Hemoglobin, Myoglobin Enzyme Action & Kinetics
04	T 02/04 R 02/06	06 05	Enzyme Action & Kinetics Protein purification
05	T 02/11 R 02/13		EXAM I Enzyme Mechanisms & Regulation
06	T 02/18 R 02/20	07 08	Enzyme Mechanisms & Regulation Lipids & Membrane structure
07	T 02/25 R 02/27	16 09	Carbohydrates Nucleic Acid Structure
08	T 03/04 R 03/06		NO CLASS SPRING BREAK NO CLASS SPRING BREAK
09	T 03/11 R 03/13	15 17	Bioenergetics Glycolysis
10	T 03/18 R 03/20		EXAM II Citric Acid Cycle (TCA Cycle)
11	T 03/25 R 03/27	20 20	Electron Transport Oxidative Phosphorylation
12	T 04/01 R 04/03	18 18	Glycogen Metabolism Gluconeogenesis
13	T 04/08 R 04/10	18 21	Pentose Phosphate Pathway Fatty Acid Metabolism
14	T 04/15 R 04/17		EXAM III Biosynthesis of Lipids
15	T 04/22 R 04/24	23 24	Nitrogen Metabolism Integration of Metabolism
16	Fri. 05/02		FINAL EXAM 01:00 PM - 03:00 PM

The instructor reserves the right to amend any or all of the constituents, requirements and policies of this syllabus at any time.